47. Total Synthesis of (\pm) -Lysergic Acid by an Intramolecular Imino-Diels-Alder Reaction

Preliminary communication

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Summary

(\pm)-Lysergic acid (1) has been synthesized from 4-hydroxymethyl-1-tosylindole (2) by a sequence of 9 steps. The crucial thermolysis $9 \rightarrow 10$ involves the *in situ*generation of the transient diene III which undergoes an intramolecular cycloaddition to a C, N-double bond at 200° and at low stationary concentration of III.

Lysergic acid, obtained by alkaline hydrolysis of ergot peptide alkaloids or more recently by isomerization of paspalic acid¹), has been shown to have structure 1 [2]. Its derivatives show a fascinating spectrum of pharmacodynamic actions which led to the established use of various valuable drugs in human therapy [3]. Previous total syntheses of (\pm) -1 have been carried out via 2,3-dihydroindole intermediates ultimately requiring a dehydrogenation step [4].

Lysergic Acid

⁶⁻Methyl-8-ergolene-8-carboxylic acid (paspalic acid), prepared by saprophytic cultivation of appropriate ergot strains in fermenters [1], as well as the C(8)-epimer of 1 (isolysergic acid) isomerize readily to lysergic acid in basic medium [1] [2].

We now report a different synthesis of (\pm) -1 which compares favorably with the former ones in terms of efficiency and length; in analogy to our recently disclosed approach to chanoclavines [5], the indole nucleus was kept intact throughout the entire reaction sequence. Our strategy (Scheme 1) relies on concomitant formation of the rings C and D by an intramolecular Diels-Alder reaction III \rightarrow IV²) followed by isomerization of the C(8), C(9)-double bond to the more stable C(9), C(10)-position¹).

To prepare the key-precursor III, several difficulties, particularly due to the instability of the 2-carbomethoxy-diene moiety, had to be resolved³). We therefore decided to start from a suitably 4-substituted indole I and to attach the diene unit in a protected form $(I \rightarrow II)$ which would not interfere with the subsequent introduction of the iminomethyl chain at position 3. Thermal liberation of the labile diene unit should permit its spontaneous interception, by the internal imino group²).

Following this plan, 4-hydroxymethyl-1-tosylindole (2) [11] was converted (2 mol-equiv. PPh₃/CBr₄, DMF, 20°, 30 min) to the crystalline bromide 3⁴) (97% yield, m.p. 133-134°) which on treatment with tributylphosphine (1.4 mol-equiv. in refluxing benzene, 2 h) gave the solid phosphonium bromide 4⁵) (100% yield, m.p. 104-106°).

Formylation of methyl bicyclo [2.2.1]hept-5-enyl-2-carboxylate [12] by successive treatment with LDA (1.1 mol-equiv., THF, -75° , 1 h) and methylformate (2.4 mol-equiv., THF, -75° , 2 h) followed by chromatography (SiO₂) furnished the *exo*-formyl compound 5^4) (68% yield) together with methyl *cis*-4, 4a, 5, 7a-tetrahydro-cyclopenta [b]pyran-3-carboxylate⁶) (15% yield). Wittig reaction of this mixture with 1 mol-equiv. of the phosphorane derived from 4 (NaCH₂S (O)CH₃, DMSO, 20°, 16 h) furnished stereoselectively the (E)-vinylindol 6^4) (62% yield, m.p. 148–150°). Removal of the N-tosyl group by heating a solution of 6 in 2 N methanolic NaOH furnished the vinylindol 7^4) (m.p. $78-79^{\circ}$, 95% yield) which underwent a Michael addition to nitroethylene [13] giving 8 in 24% yield. More efficiently (48% yield), the C(3)-functionalization $7 \rightarrow 8$ was accomplished in two steps by a Mannich reaction (HOAc, 40% aq. Me₂NH-solution, 36% aq. CH₂O-solution, 20°, 2 h) followed by treatment of the crude Mannich product with 10 mol-equiv. of nitromethane and 1.2 mol-equiv. of dimethyl acetylenedicarboxylate [14] at 20° for 5 h.

Transformation of the nitro compound 8 to the oxime ether 9 was achieved in one operation by reduction of the sodium nitronate derived from 8 with an excess solution of TiCl₃/NH₄OAc in MeOH/H₂O 3:1 at 20° for 2 h [15]

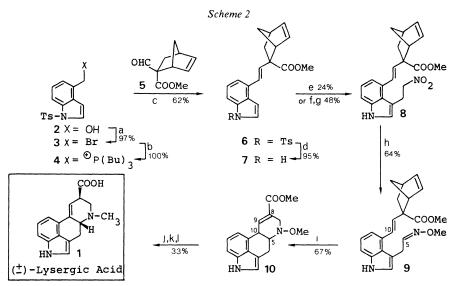
Intramolecular *Diels-Alder* reactions involving iminodienophiles are still not widely recognized [6]; for the first example see [6a]. For recent reviews on intramolecular [4+2] cycloadditions and on bimolecular imino-*Diels-Alder* reactions see [7] and [8], respectively.

^{3) 2-}Carbomethoxy-1,3-dienes dimerize spontaneously [9]. For their in situ-generation from dihydro-thiophene precursors see [10].

⁴⁾ IR.-, ¹H-NMR.- and mass spectra are in full agreement with the assigned structure.

⁵⁾ Compound 4 was characterized by IR., ¹H-NMR. and elemental analysis (C,H,N).

This side product arising from a SiO₂-promoted retro-Claisen rearrangement of the minor endoformyl isomer of 5 did not interfere with the subsequent Wittig reaction 5→6 and was fully characterized⁴) after chromatographic separation from 6.



a) CBr₄, PPh₃, DMF; b) PBu₃, C_6H_6 , 80°; c) NaH, DMSO; d) NaOH, MeOH; e) CH₂=CH-NO₂, toluene, 25°, 66 h; f) aq. CH₂O-solution, Me₂NH; g) MeNO₂, MeOOC-C \equiv C-COOMe, 20°; h) 1. NaOMe, MeOH, 2. TiCl₃/NH₄OAc, H₂N-OMe, MeQH/H₂O 3:1; 1) 200°, trichlorobenzene; J) MeOSO₂F, CH₂Cl₂, 20°; k) Al/Hg, THF/H₂O 2:1; l) 0.5 N KOH, EtOH/H₂O, under reflux.

in the presence of O-methylhydroxylamine (3 mol-equiv.). Under these reaction conditions, which permit instantaneous condensation of the labile 3-indolyl acetal-dehyde with O-methylhydroxylamine, the stable oxime ether 9^4) was obtained in 64% yield⁷) as a 2:3-mixture of syn- and anti-isomers.

We now proceeded to the crucial thermolysis of 9. It was hoped that a retro-Diels-Alder reaction would release the diene unit which might then preferentially add to the imino bond. Disappointingly, heating a 0.3% solution of 9 in bromobenzene under reflux for 1 h led quantitatively to at least three isomeric dimers of III (R = OMe). However, owing to the intramolecular nature of the desired addition, the mutually unreactive diene and dienophile should be forced to undergo cycloaddition by heating 9 to a higher temperature while keeping the stationary concentration of the transient diene III as low as possible. Accordingly, a 1% solution of 9 in 1,2,4-trichlorobenzene was added over 5 h by means of a syringe drive into preheated (200°) 1,2,4-trichlorobenzene under argon. Subsequent chromatography furnished the 8-ergolene 10^4)8 as a 2:3 mixture of diastereoisomers in 67% yield. Replacement

When the modified Nef-reaction [15] of 8 and the oximation of the resulting indolyl acetaldehyde were carried out separately 9 was obtained in only 35% yield.

⁸⁾ Compound 10 showed the following spectral data: UV. (MeOH, λ_{max} in nm (log ε)): 224 (4.18), 274 (3.81). – IR. (CH₂Cl₂, λ_{max} in cm⁻¹): 3500m, 1720s, 1450m, 1060m. – ¹H-NMR. (360 MHz, CDCl₃, 77°, internal standard tetramethylsilane (δ =0 ppm); abbreviations: s= singlet, d= doublet, t= triplet, m= multiplet, J= spin-spin coupling constant (Hz)): 2.89 ($d \times d$, J=10 and 15, 0.4 H); 3.0-3.2 (1 H); 3.2-4.1 (3.6 H); 3.66 (s, 1.2 H); 3.70 (s, 1.2 H); 3.73 (s, 1.8 H); 3.8 (s, 1.8 H); 4.1-4.4 (1 H); 6.86 (s, 0.4 H); 6.88 (s, 0.6 H); 7.00 (t, J=4, 0.4 H); 7.04-7.4 (3.6 H); 7.67 (br. s, 0.4 H); 7.84 (br. s, 0.6 H). – MS.: 299 (15, M++1), 298 (75, C₁₇H₁₈N₂O⁴₃), 267 (85), 266 (27), 208 (44), 207 (100), 180 (56). Further chromatographic fractions contained the corresponding 6-methoxy-9-ergolenes together with dimers of III (R=0Me).

of the methoxy- at N(6) by a methyl-group was accomplished by methylation of 10 with methylfluorosulfonate (3 mol-equiv., CH_2Cl_2 , 20°, 18 h) followed by hydrogenolysis of the resulting salt with amalgamated aluminium foil [16] (THF/ H_2O 2:1, 0 to +5°, 18 h). The resulting 6-methyl-ergolene mixture was heated at reflux in 0.5 N KOH in ethanol/ H_2O 1:1 under N_2 for 1 h. Concentration of the solution followed by acidification with 1 N HCl to pH 5.5 led to the precipitation of crystalline (\pm)-lysergic acid (1) (3 mg, 33% yield from 10) showing ¹H-NMR. (360 MHz in d-pyridine or in d-DMSO), IR. (nujol or KBr), UV. (0.1 N NaOH) and mass spectra identical with those of racemic lysergic acid 9) of natural origin [17].

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REFERENCES

- [1] H. Kobel, E. Schreier & J. Rutschmann, Helv. Chim. Acta 47, 1052 (1964).
- [2] L.C. Craig, Th. Shedlovsky, R.G. Gould Jr. & W.A. Jacobs, J. Biol. Chem. 125, 289 (1938);
 A. Stoll, A. Hofmann & F. Troxler, Helv. Chim. Acta 32, 506 (1949); A. Stoll, Th. Petrzilka,
 J. Rutschmann, A. Hofmann & Hs. H. Günthard, ibid. 37, 2039 (1954); P.A. Stadler & A. Hofmann,
 Helv. Chim. Acta 45, 2005 (1962).
- [3] 'Ergot Alkaloids and Related Compounds' Ed. B. Berde & H. O. Schild, Springer-Verlag 1978.
- [4] a) E. C. Kornfeld, E.J. Fornefeld, G.B. Kline, M.J. Mann, D.E. Morrison, R.G. Jones & R.B. Woodward, J. Am. Chem. Soc. 78, 3087 (1956); b) M. Julia, F. LeGoffic, J. Igolen & M. Baillargé, Tetrahedron Lett. 1969, 1569; V. W. Armstrong, S. Coulton & R. Ramage, ibid. 1976, 4311.
- [5] W. Oppolzer & J. I. Grayson, Helv. Chim. Acta 63, 1706 (1980).
- [6] a) W. Oppolzer, Angew. Chem. 84, 1108 (1972); Angew. Chem. Int. Ed. Engl. 11, 1031 (1972);
 b) S.M. Weinreb, N.A. Khatri & J. Shringarpure, J. Am. Chem. Soc. 101, 5073 (1979); B. Nader, R. W. Franck & S. M. Weinreb, ibid. 102, 1153 (1980).
- [7] W. Oppolzer, Angew. Chem. 89, 10 (1977); Angew. Chem. Int. Ed. Engl. 16, 10 (1977); W. Oppolzer, Synthesis 1978, 793; G. Brieger & J. W. Bennet, Chem. Rev. 80, 63 (1980).
- [8] S. M. Weinreb & J. I. Levin, Heterocycles 12, 949 (1979).
- [9] L.K. Sydnes, L. Skattebøl, C.B. Chapleo, D.G. Leppard, K.L. Svanholt & A.S. Dreiding, Helv. Chim. Acta 58, 2061 (1975).
- [10] J. M. McIntosh & R. A. Sieler, J. Org. Chem. 43, 4431 (1978).
- [11] G.S. Ponticello & J.J. Baldwin, J. Org. Chem. 44, 4003 (1979).
- [12] K. Seguchi, A. Sera, Y. Otsuki & K. Maruyama, Bull. Chem. Soc. Jpn. 48, 3641 (1975).
- [13] D. Ranganathan, C. B. Rao, S. Ranganathan, A. K. Mehrotra & R. Iyengar, J. Org. Chem. 45, 1185 (1980).
- [14] H. Plieninger, C. Wagner & H. Immel, Liebigs Ann. Chem. 743, 95 (1971).
- [15] J. E. McMurry & J. Melton, J. Org. Chem. 38, 4367 (1973).
- [16] G. E. Keck, S. Fleming, D. Nickell & P. Weider, Synth. Commun. 9, 281 (1979).
- [17] S. Smith & G. M. Timmis, J. Chem. Soc. 1936, 1440.
- [18] A. Stoll & A. Hofmann, Hoppe-Seylers Z. Physiol. Chem. 250, 7 (1937).

⁹⁾ Optically pure (+)- and (-)-1 have been previously obtained by resolution [18] of the racemic hydrazide of 1 [4a].